

Tetrahedron 59 (2003) 5585–5593

TETRAHEDRON

Alkynenitriles: stereoselective chelation controlled conjugate addition–alkylations

Fraser F. Fleming,* Venugopal Gudipati and Omar W. Steward

Department of Chemistry and Biochemistry, Duquesne University, Mellon Hall, Pittsburgh, PA 15282-1530, USA

Received 4 April 2003; revised 1 May 2003; accepted 14 May 2003

Abstract—Chelation-controlled conjugate addition of Grignard reagents to γ -hydroxyalkynenitriles stereoselectively generates tri- and tetra-substituted alkenenitriles. t-BuMgCl-initiated deprotonation of hydroxyalkynenitriles followed by addition of a second Grignard reagents triggers a facile conjugate addition leading to a cyclic magnesium chelate. Protonation of the chelate stereoselectively generates trisubstituted nitriles whereas the addition of t-BuLi causes conversion to an 'ate' complex that allows alkylation with aldehyde electrophiles. The chelation-controlled conjugate addition–alkylation generates tri- and tetra-substituted alkenenitriles that are otherwise difficult to synthesize. $©$ 2003 Elsevier Science Ltd. All rights reserved.

1. Introduction

Chelation provides a powerful means of bond construction.^{[1](#page-7-0)} Temporary chelation dramatically accelerates many reactions by virtue of the close proximity imparted on the two reactive centers, essentially harnessing the inherent entropic advantages of intramolecular reactions^{[2](#page-7-0)} in a formal intermolecular process. Tethering the two reacting centers generally imposes precise geometric constraints on the transition state resulting in high stereoselectivity during bond formation,^{[3](#page-7-0)} in some cases affording stereoisomers that are otherwise difficult to access.[4](#page-7-0)

Conjugate addition^{[5](#page-7-0)} and carbometallation^{[6](#page-7-0)} reactions, especially, are dramatically enhanced by chelation. The enhanced reactivity is readily apparent by comparing chelation-controlled^{[7](#page-7-0)} and organocopper^{[8](#page-7-0)} conjugate additions to alkenenitriles—substrates that are generally recalcitrant acceptors in anionic conjugate addition reactions.[9](#page-7-0) For example, sequential deprotonation of 1 with t-BuMgCl (Scheme 1) and alkyl exchange from a variety of Grignard reagents triggers a stereoselective conjugate

Scheme 1. Chelation-controlled conjugate addition.

addition through chelate 2 whereas cuprates are unreactive toward cyclohexenecarbonitrile $(1, HO=H)$.

The enhanced reactivity, and high stereoselectivity, of chelation-controlled conjugate additions to alkenenitriles suggested extending the strategy to alkynenitriles. The attraction lies not only in enlarging chelation-controlled conjugate additions to incorporate sp-hybridized $\frac{10}{2}$ $\frac{10}{2}$ $\frac{10}{2}$ but in stereoselectively assembling tri-substituted alkenenitriles (Scheme 2, $R^3 = H$).^{[11](#page-7-0)} Conceptually, chelation-controlled conjugate addition, followed by intercepting the intermediate vinyl organometallic 6 with carbon electrophiles,¹² provides a concise route to tetra-substituted alkenenitriles $\overline{7}$ for which few syntheses exist.¹³ This account describes the scope of chelation-controlled con-jugate additions to hydroxyalkynenitriles^{[14](#page-7-0)} and the first series of conjugate addition–alkylation reactions with alkynenitriles. Collectively these chelation-controlled conjugate addition reactions stereoselectively assemble triand tetra-substituted alkenenitriles that are otherwise challenging synthetic targets.

Scheme 2. Chelation-controlled conjugate addition–alkylation of alkynenitriles.

Keywords: conjugate addition; alkynenitriles.

^{*} Corresponding author. Tel.: $+1-412-396-6031$; fax: $+1-412-396-5683$; e-mail: flemingf@duq.edu

Table 1. Hydroxyalkynenitrile synthesis

2. Results and discussion

A prerequisite for chelation-controlled conjugate addition reactions is an efficient synthesis of diverse hydroxyalkynenitriles. A general route to hydroxyalkynenitriles must overcome the instability of the parent 4-hydroxybutynenitrile $(4a)$ toward base^{[15](#page-7-0)} and silica gel chromatography[16](#page-7-0) that stems partly from the enhanced electrophilicity imparted by the proximal hydroxyl group, 17 and partly from the accessibility of the electrophilic β -carbon. Moderating the reactivity by THP protec-tion of the hydroxyl group^{[18](#page-7-0)} proved admirable in allowing a high yielding deprotonation–cyanation^{[19](#page-7-0)} route to the chromatographically stable THP-protected alkynenitrile that was subsequently converted to spectroscopically pure 4a on exposure to Dowex 50-X4 (Table 1, entry 1).

Table 2. Conjugate additions to alkynenitriles

Analogous lithiation–cyanations of alkynes 8b and 8c generate the chromatographically stable alkynenitriles 4b and $4c$ (Table 1, entries $2-3$), with the sterically hindered nitrile 4c imparting sufficient stability for direct double lithiation–cyanation of the hydroxyalkyne 8c (Table 1, entry 3).

Hydroxyalkynenitriles 4a and 4c react with a range of Grignard reagents in an efficient chelation-controlled conjugate addition (Table 2). Experimentally, t-BuMgCl deprotonates the γ -hydroxyalkynenitriles permitting conjugate addition upon addition of a slight excess of a second. potentially more valuable, Grignard reagent. t-BuMgCl is an excellent sacrificial base for the initial deprotonation 20 since no transfer of the t-butyl group occurs during deprotonation at -78° C, whereas warming to ambient temperature with excess t -BuMgCl efficiently induces conjugate addition of the t-butyl group (Table 2, entry 4). High steric demand is tolerated in the Grignard reagent and in the alkynenitrile, although the reaction efficiency is reduced relative to reactions with less-substituted counterparts (compare Table 2, entry 1 with entries 4 and 9; entry 1 with entry 2; and entry 5 with entry 6).

Chelation-controlled conjugate additions of alkyl, vinyl, aryl, and alkynyl Grignards exclusively generate E-alkenenitriles. 21 Modest halogen, silicon, and sulfur functionality is tolerated in the Grignard reagent (Table 2, entries $10-12$) although the instability^{[22](#page-7-0)} of the Grignard prepared from 3-chloropropyl phenylsulfide^{[23](#page-7-0)} (Table 2, entry 11) causes concomitant formation of 7l, presumably through conjugate addition of PhSMgCl formed by cyclization of the Grignard reagent.

Table 2 α

Chelate 6 is surprisingly unreactive toward alkylation even being unreactive toward benzaldehyde! 24 The low nucleophicity of chelate 6 is consistent with the inability of 6 to act as a vinyl Grignard reagent in a second chelationcontrolled conjugate addition with alkynenitrile 9a ([Scheme 3\)](#page-3-0). Competitive conjugate addition of the chelate 6 is only observed with $Me₃SiCH₂MgCl$ [\(Table 1](#page-1-0), entry 12), presumably reflecting formation of the silicate $11n^{25}$ $11n^{25}$ $11n^{25}$ from 10n which unmasks a more reactive $C-Mg$ bond^{[26](#page-7-0)} that competes with the hindered $Me₃SiCH₂MgCl$ for **9a**.

Mechanistically, the stepwise chelation controlled conjugate addition directly parallels the analogous reactions of hydroxyalkenenitriles.^{[7](#page-7-0)} t-BuMgCl-induced deprotonation of the γ -hydroxyalkynenitrile followed by halogen–alkyl exchange^{[5b](#page-7-0)} [\(Scheme 3\)](#page-3-0), leads to the alkylmagnesium

alkoxide 5 that triggers the key anionic conjugate addition. Alkyl transfer from 5 requires overlap of the σ carbon– magnesium bond with the π^* -LUMO where the π^* -node precludes a concerted addition^{[27](#page-7-0)} as does the large distance between the magnesium and α -carbon atoms. Alkyl transfer from 5, with activation of the nitrile by MgX_2 , leads directly to the magnesiated alkenenitrile 10 that will equilibrate^{[28](#page-7-0)} to the cyclic chelate $6.^{29}$ $6.^{29}$ $6.^{29}$ Subsequent protonation of 6 generates the corresponding alkenenitrile 7 with retention of stereo-chemistry.^{[6](#page-7-0)}

Chelation is essential for the conjugate addition. A control experiment in which the THP-protected alkynenitrile 4a $HO=OTHP$ is exposed to *n*-BuMgCl leads to 90% recovery of unreacted nitrile. Similarly, positioning the hydroxyl group two atoms removed from the alkynenitrile prevents

Scheme 3. Chelation-controlled conjugate addition mechanism.

the conjugate addition. The inability of alkynenitrile 4b to undergo chelation controlled conjugate addition is surprising given the related carbometallation of butynols with Grignard reagents, 6 and the apparently more favorable proximity achieved by increasing the length of the tether $(Eq. (1))$. Although speculative, the inability of a chelationcontrolled conjugate addition with 4b may stem from an interaction^{[30](#page-7-0)} between magnesium and the π -bond^{[31](#page-7-0)} which positions the alkyl group away from the alkyne and deactivates the alkynenitrile nitrogen toward complexation with magnesium dihalide.

$$
\begin{array}{c}\n\begin{array}{c}\n\hline\n\end{array} \\
\begin{array}{c}\n\hline\n\end{array} \\
\begin{array}{c}\n\hline\n\end
$$

The poor nucleophilicity of the chelate 6 is surprising for a formal 'dianion.' Efforts to activate the chelate 6 by weakening the Mg-sp² bond focused on the addition of t-BuLi for conversion to the more reactive magnesium ate species 13 (Scheme 4).^{[32](#page-7-0)} Although not proof for the intermediacy of an ate complex, the addition of t-BuLi sufficiently activates 6 for a stereoselective^{[33](#page-7-0)} alkylation with aldehydes 34 that is consistent with a retentive

alkylation^{[6a](#page-7-0)} through a cyclic ate complex $(Table 3)$. The overall sequence of t-BuMgBr deprotonation, RMgX conjugate addition, t-BuLi ate activation, and alkylation is remarkably efficient for several Grignard reagents and even with the potentially enolizable aldehyde 3-phenylpropanal (Table 3, entries 3–6). Collectively, the three component coupling stereoselectively generates tetrasubstituted alkenenitriles that are otherwise difficult to synthesize.

Scheme 4. Chelation-controlled conjugate addition–alkylations.

Temporary chelation of Grignard reagents to γ -hydroxyalkynenitriles triggers a facile conjugate addition reaction. t -BuMgCl-initiated deprotonation of γ -hydroxyalkynenitriles and addition of a slight excess of a second Grignard reagent causes a stepwise conjugate addition resulting in the formation of a cyclic magnesium chelate. Protonation of the chelate stereoselectively generates tri-substituted alkenenitriles whereas addition of t-BuLi generates a more reactive ate complex that alkylates aromatic and aliphatic aldehydes with complete stereochemical fidelity. Collectively the chelation-controlled conjugate addition–alkylation generates a range of tri- and tetra-substituted alkenenitriles that are otherwise difficult to synthesize.

3. Experimental 35

3.1. Data for compounds

3.1.1. 4-(Tetrahydro-pyran-2-yloxy)-but-2-ynenitrile. A hexanes solution of n-BuLi (5.4 mL, 8.6 mmol) was added to a -78° C, THF solution of $8a^{18}$ $8a^{18}$ $8a^{18}$ (1.2 g, 8.6 mmol) followed, after 15 min, by a THF solution of $PhOCN¹⁹$ $PhOCN¹⁹$ $PhOCN¹⁹$ (1.3 equiv.). The cooling bath was removed and, after 15 min, aqueous NaOH (6 M) was added and followed by ether extraction. The combined extracts were washed sequentially with NaOH (6 M), and saturated NaCl, passed through a short plug of silica gel $(5\times1$ cm column), dried over NaSO4, concentrated, and purified by radial chromatography (1:9 EtOAc/hexanes) to yield 1.3 g (92%) of 4-(tetrahydro-pyran-2-yloxy)-but-2-ynenitrile as oil: IR (film) 2304, 2265 cm⁻¹; ¹H NMR 1.52-1.84 (m, 6H), $3.54 - 3.58$ (m, 1H), $3.75 - 3.83$ (m, 1H), 4.37 (s, 2H), 4.76 (s, 1H); 13C NMR ^d 18.6, 25.0, 29.9, 53.6, 59.7, 62.0, 81.6, 97.7, 104.6; MS m/e 164 (M-H).

3.1.2. 4-Hydroxy-but-2-ynenitrile (4a). An anhydrous methanolic solution (15 mL) of 5-(tetrahydro-pyran-2 yloxy)-pent-2-ynenitrile (0.6 g) and Dowex (pre washed with anhydrous methanol) was stirred at room temperature for 1.5 h. The reaction was then filtered, the residue concentrated, and retreated with Dowex for 1.5 h to yield 295 mg (100%) of 4a as oil: IR (film) 3447, 2308, 2245 cm⁻¹; ¹H NMR δ 2.39 (br, 1H), 4.40 (s, 2H), 2.38 $(s, 1H)$; ¹³C NMR δ 50.6, 59.6, 83.1, 104.6.

3.1.3. 5-(Tetrahydro-pyran-2-yloxy)-pent-2-ynenitrile. A hexanes solution (1.6 M) of *n*-BuLi $(0.52 \text{ mL}, 0.84 \text{ mmol})$ was added to a -78° C, THF solution of $8b^{36}$ $8b^{36}$ $8b^{36}$ (0.13 g, 0.84 mmol) followed, after 15 min, by a THF solution of PhOCN (1.3 equiv.) . The cooling bath was removed and, after 15 min, aqueous NaOH (6 M) was added and followed by ether extraction. The combined extracts were washed sequentially with NaOH (6 M), and saturated NaCl, passed through a short plug of silica gel $(5\times1$ cm column), dried over NaSO4, concentrated and, purified by radial chromatography $(1:10 \text{ EtoAc/hexanes})$ to yield 0.12 g (80%) of 5-(tetrahydro-pyran-2-yloxy)-pent-2-ynenitrile as oil: IR (film) 2314, 2261 cm⁻¹; ¹H NMR δ 1.57-1.84 (m, 6H), 2.65–2.69 (m, 2H), 3.52–3.65 (m, 2H), 3.81–3.92 (m, 2H), 4.64 (brs, 1H); 13C NMR ^d 19.2, 20.6, 25.3, 30.4, 55.9, 62.3, 63.7, 84.7, 99.0, 105.1; MS e/m 178 (M-H).

3.1.4. 5-Hydroxy-pent-2-ynenitrile (4b). A methanolic solution (10 mL) of 5-(Tetrahydro-pyran-2-yloxy)-pent-2 ynenitrile (0.12 g) and Dowex (pre-washed with anhydrous methanol) was stirred at room temperature for 1.5 h. The reaction was then filtered, the residue concentrated, and retreated with Dowex for 1.5 h to yield 63 mg (100%) of **4b** as an oil: IR (film) 3434, 2315, 2263 cm⁻¹; ¹H NMR δ 2.12 (s, 1H), 2.65–2.61 (s, J=6.0 Hz, 2H), 3.82 (s,

2H); 13C NMR ^d 23.0, 56.1, 59.3, 85.0, 105.0 MS m/e 95 $(M+H)$.

3.1.5. 4-Hydroxy-4-methyl-pent-2ynenitrile (4c). A hexanes solution $(1.6 M)$ of *n*-BuLi $(2.96 mL, 4.75 mmol)$ was added to a -78° C, THF solution of 8c (0.4 g, 4.75 mmol) followed, after 15 min, by a THF solution of PhOCN (1.3 equiv.). The cooling bath was removed and, after 15 min, aqueous NaOH (6 M) was added and followed by ether extraction. The combined extracts were washed sequentially with NaOH (6 M), and saturated NaCl, dried over NaSO4, concentrated and, purified by radial chromatography (1:9 EtOAc/hexanes) to yield 0.41 g (80%) of 4c as oil, identical to material previously synthesized.[37](#page-8-0)

3.2. General conjugate addition procedure

A THF solution of t -BuMgCl (1.0 equiv., $1-2$ M) was added to a -78° C, THF solution of the γ -hydroxyalkynenitrile (1 equiv.) followed, after 5 min, by a THF solution of the appropriate Grignard reagent $(1.1 \text{ equiv.}, 1-3 \text{ M})$. After 45 min, the reaction mixture was allowed to warm to room temperature (15 min) and then saturated aqueous $NH₄Cl$ was added. The crude reaction mixture was extracted with EtOAc the combined organic extracts were dried $(Na₂SO₄)$, passed through a short plug of silica gel $(2\times1$ cm column), concentrated, and purified by radial chromatography.

3.2.1. 4-Hydroxy-3-but-2-enenitrile (7a). Performing the general conjugate addition procedure with a THF solution (5 mL) of 4a (20 mg) and MeMgCl provided, after purification by radial chromatography (1:4 EtOAc/ hexanes), 22 mg (92%) of 7a spectrally identical to material previously synthesized.[21a](#page-7-0)

3.2.2. (2E)-4-Hydroxy-3,4-dimethylpent-2-enenitrile (7b). Performing the general conjugate addition procedure with a THF solution (5 mL) of 4c (20 mg) and MeMgCl provided, after purification by radial chromatography (1:4 EtOAc/hexanes), 16.1 mg $(70%)$ of **7b** as an oil: IR (film) 3410, 2932, 2215, 1601 cm⁻¹; ¹H NMR δ 1.37 (s, 6H), 2.08 $(s, 3H), 5.63$ $(s, 1H);$ ¹³C NMR δ 17.8, 28.5, 73.6, 94.1, 117.5, 169.6; MS e/m 126 (M+H).

3.2.3. (2E)-3-(Hydroxymethyl)-4-methylpent-2-enenitrile (7c). Performing the general conjugate addition procedure with a THF solution (5 mL) of 4a (12.9 mg) and i-PrMgBr provided, after purification by radial chromatography (3:7 EtOAc/hexanes), 17.3 mg (87%) of $7c$ as an oil: IR (film) 3447, 2221, 1628 cm⁻¹; ¹H NMR δ 1.16 (d, $J=7.2$ Hz, 6H), 1.21 (br s, 1H), 3.10 (sept, $J=7$ Hz, 1H), 4.30 (s, 2H), 5.53 (s, 1H); ¹³C NMR δ 20.5, 32.2, 61.0, 91.9, 116.9, 172.1; MS m/e 126 (M+H).

3.2.4. (2E)-3-(Hydroxymethyl)-4,4-dimethylpent-2-enenitrile (7d). Performing the general conjugate addition procedure with a THF solution (5 mL) of 4a (16.2 mg) and t-BuMgCl provided, after purification by radial chromatography (1:4 EtOAc/hexanes), 16.5 mg (60%) of $7d$ as an oil: IR (film) 3483, 2219, 1636 cm⁻¹; ¹H NMR δ 1.32 (s, 9H), 1.75 (s, 1H), 4.32 (s, 2H), 5.72 (s, 1H); ¹³C NMR δ 28.9, 36.2, 62.4, 91.4, 118.0, 173.2; MS m/e 140 (M+H).

3.2.5. $(2E)$ -4-Hydroxy-3-phenylbut-2-enenitrile $(7e)$. Performing the general conjugate addition procedure with a THF solution (5 mL) of 4a (15 mg) and PhMgCl provided, after purification by radial chromatography (3:7 EtOAc/ hexanes), 27 mg (92%) of **7e** as an oil:^{[38](#page-8-0)} IR (Film) 3446, 3060, 2221, 1623 cm⁻¹;¹H NMR δ 2.37 (br s, 1H), 4.52 (s, 2H), 5.78 $(s, 1H), 7.43$ $(s, 5H);$ ¹³C NMR δ 64.6, 94.2, 117.4, 127.2, 128.9, 130.1, 134.7, 163.6; MS m/e 159.

 $3.2.6.$ $(2E)$ -4-Hydroxy-4-methyl-3-phenylpent-2-enenitrile (7f). Performing the general conjugate addition procedure with a THF solution (5 mL) of $4c$ (20 mg) and PhMgCl, provided after purification by radial chromatography (1:4 EtOAc/hexanes), 22.6 mg (66%) of $7f$ as a light brown solid: IR (film) 3440, 3060, 2229, 1614 cm⁻¹; ¹H NMR δ 1.39 (s, 6H), 1.86 (s, 1H), 5.97 (s, 1H), 7.19-7.44 (m, 5H); ¹³C NMR δ 28.7, 73.6, 97.1, 116.9, 127.7, 128.4, 128.7; HRMS (ESI) calcd for $(M+Na^+)$ C₁₂H₁₃NONa⁺ 210.0889, found 210.08933.

3.2.7. (2E)-3-(Hydroxymethyl) penta-2, 4-dienenitrile (7g). Performing the general conjugate addition procedure with a THF solution (5 mL) of $4a$ (21.3 mg) and vinyl magnesium bromide provided, after purification by radial chromatography (1:4 EtOAc/hexanes), 25 mg (87%) of $7g$ as an oil: IR (film) 3423, 2218, 1635, 1582 cm⁻¹, ¹H NMR δ 2.08 (s, 1H), 4.51 (s, 2H), 5.54 (d, $J=11$ Hz, 1H), 5.58 (d, $J=17$ Hz, 1H), 5.69 (s, 1H), 6.87 (dd, $J=17$, 11 Hz, 1H); ¹³C NMR: 61.0, 95.7, 116.6, 120.9, 131.4, 158.0; MS m/e 109.

3.2.8. (2E)-3-(Hydroxymethyl)-5-phenylpent-2-en-4-ynenitrile (7h). Performing the general conjugate addition procedure with a THF solution (5 mL) of 4a (30 mg) and $PhC \equiv CMgCl³⁹$ provided, after purification by radial chromatography (1:3 EtOAc/hexanes), 57.6 (85%) mg of **7h** as an oil: IR (film) 3347, 3067, 2224, 2192, 1589 cm⁻¹;
¹H NMR δ 2.36 (s. 1H) 4.39 (s. 2H) 5.93 (s. 2H) 7.36-¹H NMR δ 2.36 (s, 1H), 4.39 (s, 2H), 5.93 (s, 2H), 7.36– 8.74 (m, 5H); 13C NMR: 64.1, 83.2, 100.9, 102, 116.7, 121.2, 128.5, 129.9, 132.3, 145.8 MS m/e 183 (M⁺).

3.2.9. (2Z)-4,4-Dibutyl-3-(hydroxymethyl)-4-stannaoct-2-enenitrile (7i). Performing the general conjugate addition procedure with a THF solution (5 mL) of $4a(10 \text{ mg})$ and (Bu) ₃SnMgCl^{[40](#page-8-0)} provided, after purification by radial chromatography (1:4 EtOAc/hexanes), 32 mg (70%) of $7i$ as oil: IR (film) 3422, 2217, 1654 cm⁻¹; ¹H NMR δ 0.91 (t, $J=7.0$ Hz, 9H), $1.12-1.64$ (m, 18H), 4.42 (s, 2H), 6.26 (s, 1H); 13C NMR ^d 10.1, 13.6, 27.2, 29.0, 68.1, 105.5, 118.9, 176.7; MS m/e 373 (M⁺).

3.2.10. (2E)-7-Chloro-3-(hydroxymethyl)-hept-2-enenitrile (7j). Performing the general conjugate addition procedure with a THF solution (5 mL) of $4a(15 \text{ mg})$ and chlorobutyl-magnesium bromide^{[41](#page-8-0)} provided, after purification by radial chromatography (3:7 EtOAc/hexanes), 25 mg (78%) of 7*j* as an oil: IR (Film) 3435, 2220, 1639 cm⁻¹, ¹H NMR δ 1.63- 1.73 (m, 2H), $1.80 - 1.89$ (m, 2H), 2.42 (t, $J=7.5$ Hz, 2H), 3.57 $(t, J=6.0 \text{ Hz}, 2\text{H})$, 4.26 (s, 2H), 5.59 (s, 1H); ¹³C NMR: 25.3, $31.2, 31.6, 44.3, 63.9, 94.3, 116.8, 166.2; MS *m/e* 173 (M+H).$

3.2.11. (2E)-3-(Hydroxymethyl)-6-phenylthiohex-2-enenitrile (7k) and (2Z)-4-hydroxy-3-phenylthiobut-2-enenitrile (7l). Performing the general conjugate addition

procedure with a THF solution (5 mL) of $4a$ (16 mg) and $PhS(CH_2)_3MgCl^{22}$ $PhS(CH_2)_3MgCl^{22}$ $PhS(CH_2)_3MgCl^{22}$ provided, after purification by radial chromatography (3:7 EtOAc/hexanes), 19.3 mg (42%) of $7k$ and 20 mg (53%) of 7l as an oils. For 7k IR (film) 3426, 3060, 2219, 1632, 1584 cm⁻¹; ¹H NMR δ 1.60 (br s, 1H), $1.78-1.88$ (m, 2H), 2.51 (dd, $J=8$ Hz, 2H), 2.95 (t, $J=7.3$ Hz, 2H), 4.20 (s, 2H), 5.57 (s, 1H), 7.17–7.60 (m, 5H); 13C NMR ^d 27.6, 31.0, 33.5, 64.0, 94.6, 126.4, 129.0, 129.8, 165.7; MS e/m 233. For 7l IR (film) 3438, 3058, 2216, 1579 cm⁻¹; ¹H NMR δ 2.03 (s, 1H), 4.07 (s, 2H), 5.81 $(s, 1H), 7.27 - 7.55$ (m, 5H); ¹³C NMR δ 64.1, 93.6, 127.9, 129.9, 134.7, 162; MS m/e 191 (M⁺).

3.2.12. (2Z)-3-(Hydroxymethyl)-5,5-dimethyl-5-silahex-2-enenitrile $(7m)$ and $(2E)$ -1-[1-(2,2-dimethyl-2-silapropyl)-2-hydroxyethylidene]-2-(hydroxymethyl)prop-2-ene-1,3-dicarbonitrile (7n). Performing the general conjugate addition procedure with a THF solution (5 mL) of $4a$ (20 mg) and (CH₃)₃SiCH₂MgCl provided, after purification by radial chromatography (3:7 EtOAc/ hexanes), 16 mg (38.3%) of $7m$ and 14 mg (19.4) of $7n$ as an oils. For 7m: IR (Film) 3456, 2215, 1620; ¹H NMR δ 0.12 (s, 9H), 1.94 (s, 2H), 4.13 (s, 2H), 5.39 (s, 1H); 13C NMR δ -1.0, 24.6, 65.3, 89.2, 118.1, 166.7; MS m/e 169. For **7n**: IR (Film) 3456, 2221, 1630 cm⁻¹; ¹H NMR δ 0.22 (s, 9H), 2.38 (s, 2H), 4.27 (s, 2H), 4.36 (s, 2H), 5.84 (s, 1H); ¹³C NMR δ-0.8, 28.0, 62.4, 63.3, 100.2, 100.6, 116.3, 157.1, 167.5; MS m/e 250 (M+H).

3.3. General conjugate addition–alkylation procedure

A THF solution of t -BuMgCl (1.0 equiv., $1-2$ M) was added to a -78° C. THF solution of the γ -hydroxyalkynenitrile (1 equiv.) followed, after 5 min, by a THF solution of the appropriate Grignard reagent $(1.1 \text{ equiv.}, 1-3 \text{ M})$. After 45 min, the reaction mixture was allowed to warm to room temperature (15 min), re-cooled to -78° C, and then a hexanes solution of t -BuLi (1.2 equiv., 1.5 M) was added. The cooling bath was then removed, and after 15 min neat aldehyde (1.5 equiv.) was added, followed, after a further 30 min, by aqueous saturated NH4Cl. The crude reaction mixture was extracted with EtOAc, the combined extracts were passed through a short plug of silica gel $(2\times1$ cm column), concentrated, and purified by radial chromatography.

3.3.1. (2Z)-4-Hydroxy-2-(hydroxyphenylmethyl)-3 phenylbut-2-enenitrile (7o). Performing the general conjugate addition–alkylation procedure with a THF solution (5 mL) of 4a (10 mg), PhMgCl and PhCHO provided, after purification by radial chromatography (3:7 EtOAc/ hexanes), 19.6 mg (60%) of 7σ as a light brown solid (mp) 123-125°C): IR (film) 3388, 2219, 1595 cm⁻¹; ¹H NMR δ 4.55–4.57 (m, 2H), 5.40–5.42 (m, 1H, exchanges with D₂O), 5.94–5.96 (m, 1H), 6.21 (d, J=4.0 Hz, 1H, exchanges with D₂O), 7.26–7.50 (m, 10H); ¹³C NMR δ 60.6, 67.6, 117.4, 118.2, 125.9, 127.3, 128.1, 138.5, 141.6, 156.0 MS m/e 247 (M-H₂O).

3.3.2. (2Z)-3-(Hydroxymethyl)-2-(hydroxyphenylmethyl)-4-methylpent-2-enenitrile (7p). Performing the general conjugate addition–alkylation procedure with a THF solution (5 mL) of $4a$ (20 mg) , *i*-PrMgCl and PhCHO provided, after purification by radial chromatography (4:6 EtOAc/hexanes), 41.1 mg (72%) of 7p as an oil: IR (Film) 3414, 2217, 1603 cm⁻¹; ¹H NMR δ 1.09 (d, J=6.7 Hz, 3H), 1.15 (d, J=7.1 Hz, 3H), 2.64 (s, 1H), 3.13–3.22 (m, 1H), 3.45 $(s, 1H), 4.30 (ABq, \Delta \nu=31.5 Hz, J=12.3 Hz, 2H), 5.79 (s, 1H),$ $7.32 - 7.45$ (m, 5H); ¹³C NMR δ 20.2, 20.3, 35.2, 57.0, 70.0, 116.5, 117.0 126.0, 128.7, 140.1, 163.4 MS m/e 231 (M-H).

3.3.3. (2Z)-3-(Hydroxymethyl)-2-(1-hydroxy-3-phenylpropyl)-4-methylpent-2-enenitrile (7q). Performing the general conjugate addition–alkylation procedure with a THF solution (5 mL) of $4a$ (12 mg) , *i*-PrMgCl and 3-phenylpropanal provided, after purification by radial chromatography (3:7 EtOAc/hexanes), 24.9 mg (65%) of 7q as an oil: IR (film) 3410, 2214, 1603 cm⁻¹; ¹H NMR δ 1.07 (d, J=6.4 Hz, 3H), 1.15 (d, J=6.3 Hz, 3H), 1.70 (br s, 1H), $1.97 - 2.21$ (m, 2H), 2.32 (br s, 1H), 2.74 (t, J=7.5 Hz, $2H$), $3.14-3.23$ (m, 1H), 4.14 (s, $2H$), 4.55 (t, $J=6.8$ Hz, 1H), 7.12–7.31 (m, 5H); 13C NMR ^d 20.2, 20.4, 31.8, 35.1, 36.8, 57.0, 67.4, 116.3, 117.1, 126.2, 128.4, 128.5, 140.8, 164.3.

3.3.4. (2E)-4-Hydroxy-2-(1-hydroxy-3-phenylpropyl)-3 phenylbut-2-enenitrile (7r). Performing the general conjugate addition–alkylation procedure with a THF solution (5 mL) of 4a (15 mg), PhMgCl and 3-phenylpropanal provided, after purification by radial chromatography (3:7 EtOAc/hexanes), 30.9 mg $(57%)$ of $7r$ as an oil. IR (film) 3386, 2216, 1602 cm⁻¹; ¹H NMR δ 2.1-2.24 (m, 2H), 2.74–2.78 (m, 2H), 3.20 (brs, 2H), 4.39 (AB_q, $\Delta \nu$ =40.0 Hz, $J=13.4$ Hz, 2H), 4.69 (t, $J=7$ Hz, 1H), 7.20–7.38 (m, 10H); 13C NMR: 31.7, 37.1, 62.3, 67.6, 116.9, 118.2, 126.3, 127.8, 128.4, 128.6, 128.8, 129.6, 137.5, 140.6, 157.4.

3.3.5. (2Z)-4-Hydroxy-2-(1-hydroxy-3-phenylpropyl)-4 methyl-3-phenylpent-2-enenitrile (7s). Performing the general conjugate addition–alkylation procedure with a THF solution (5 mL) of 4c (17 mg), PhMgCl and 3-phenylpropanal provided, after purification by radial chromatography $(3.7 \text{ EtOAc/hexanes})$, 28.5 mg $(57%)$ of 7s as an oil; IR (film) 3434, 2215, 1602 cm⁻¹; ¹H NMR δ 1.25 (s, 3H), 1.36 (s, 3H), 1.40–1.60 (m, 2H), 2.01–2.27 (m, 2H), 2.72– 2.92 (s, 2H), $5.22 - 5.27$ (m, 1H), $7.20 - 7.71$ (m, 10H); 13 C NMR: 30.6, 31.4, 32.1, 37.2, 67.1, 74.9, 116.8, 120.1, 126.1, 127.1, 127.9, 128.5, 139.4, 141.2, 165.1; HRMS (ESI) calcd for $(M+Na^+)$ C₂₁H₂₃NO₂ 344.1621, found 344.1611.

3.3.6. (2Z)-4-Hydroxy-2-(1-hydroxy-3-phenylpropyl)- 3,4-dimethylpent-2-enenitrile (7t). Performing the general conjugate addition–alkylation procedure with a THF solution (5 mL) of $4c$ (20 mg) , MeMgCl and 3-phenylpropanal, provided after purification by radial chromatography $(3:7 \text{ EtOAc/hexanes})$, 29 mg (61%) of 7t as an oil: IR (film) 3433, 2211, 1602 cm⁻¹; ¹H NMR δ 1.21-1.37 (m, 1H), 1.34 (s, 3H), 1.42 (s, 3H), 1.94–2.20 (m, 2H), 2.09 (s, 3H), 2.65–2.85 (m, 3H), 5.07–5.12 (m, 1H), 7.18–7.46 (m, 5H); 13C NMR: 22.0, 29.7, 30.3, 32.0, 37.3, 66.8, 75.2, 116.5, 117.7, 125.9, 128.4 (doubled), 141.3, 161.8; MS m/e 259 (M-OH).

Acknowledgements

Financial support of this research from NIH, and travel funds from NSF, are gratefully acknowledged.

References

- 1. Hoveyda, A. H.; Evans, D. A.; Fu, G. C. Chem. Rev. 1993, 93, 1307.
- 2. Eliel, E. L.; Wilen, S. H.; Mander, L. N. Stereochemistry of Organic Compounds; Wiley: New York, 1994; pp 676–682.
- 3. Mikami, K.; Shimizu, M.; Zhang, H.-C.; Maryanoff, B. E. Tetrahedron 2001, 57, 2917.
- 4. (a) White, J. D. Pure Appl. Chem. 1994, 66, 2183. (b) White, J. D.; Shin, H.; Kim, T.-S.; Cutshall, N. S. J. Am. Chem. Soc. 1997, 119, 2404.
- 5. For pioneering chelation-controlled additions to enones see: (a) Swiss, K. A.; Hinkley, W.; Maryanoff, C. A.; Liotta, D. C. Synthesis 1992, 127. (b) Swiss, K. A.; Liotta, D. C.; Maryanoff, C. A. J. Am. Chem. Soc. 1990, 112, 9393. (c) Solomon, M.; Jamison, W. C. L.; McCormick, M.; Liotta, D.; Cherry, D. A.; Mills, J. E.; Shah, R. D.; Rodgers, J. D.; Maryanoff, C. A. J. Am. Chem. Soc. 1988, 110, 3702.
- 6. (a) Fallis, A. G.; Forgione, P. Tetrahedron 2001, 57, 5899. (b) Marek, I.; Normant, J. F. In Metal Catalyzed Cross Coupling Reactions; Diederich, F., Stang, P. J., Eds.; Wiley-VCH: Weinheim, 1998; pp 271–337. (c) Normant, J. F.; Alexakis, A. Synthesis 1981, 841.
- 7. Fleming, F. F.; Wang, Q.; Zhang, Z.; Steward, O. W. J. Org. Chem. 2002, 67, 5953.
- 8. House, H. O.; Umen, M. J. J. Org. Chem. 1973, 38, 3893.
- 9. Fleming, F. F.; Wang, Q. Chem. Rev. 2003, 103, 2035.
- 10. Chelation-controlled conjugate additions to activated alkynes appear not to have been previously examined.¹
- 11. Kiefel, M. J. Comprehensive Organic Functional Group Transformations; Katritzky, A. R., Meth-Cohn, O., Rees, C. W., Eds.; Pergamon: Cambridge, 1995; Vol. 3, pp 641–676.
- 12. For conjugate addition–protonation of alkynenitriles see: (a) Westmijze, H.; Kleijn, H.; Vermeer, P. Synthesis 1978, 454. (b) Kleijn, H.; Westmijze, H.; Meijer, J.; Vermeer, P. J. Organomet. Chem. 1981, 206, 257. (c) Westmijze, H.; Kleijn, H.; Vermeer, P. Tetrahedron Lett. 1979, 3327. For conjugate addition–alkylations of organocopper reagents to carbonyl-activated alkynes see: (d) Kennedy, J. W. J.; Hall, D. G. J. Am. Chem. Soc. 2002, 124, 898. (e) Wei, H.-X.; Steven, W.; Li, G. Synth. Commun. 1999, 29, 2959. (f) Wei, H.-X.; Hook, J. D.; Fitzgerald, K. A.; Li, G. Tetrahedron: Assymetry 1999, 10, 661. (g) Li, G.; Wei, H.-X.; Whittlesey, B. R.; Batrice, N. N. J. Org. Chem. 1999, 64, 1061. (h) Wei, H.-X.; Willis, S.; Li, G. Tetrahedron Lett. 1998, 39, 8203. (i) Hall, D. G.; Deslongchamps, P. J. Org. Chem. 1995, 60, 7796. (j) Hall, D. G.; Chapdelaine, D.; Preville, P.; Deslongchamps, P. Synlett 1994, 660. (k) Mongrain, C.; Gaudreault, R. C. Synth. Commun. 1990, 20, 2491. (l) Corriu, R. J. P.; Moreau, J. J. E.; Vernhet, C. Tetrahedron Lett. 1987, 28, 2963. (m) Carlson, R. M.; Oyler, A. R.; Peterson, J. R. J. Org. Chem. 1975, 40, 1610.
- 13. (a) Ono, N.; Tamura, R.; Nakatsuka, T.; Hayami, J.; Kaji, A. Bull. Chem. Soc. J. 1980, 53, 3295. (b) Ojima, I.; Kumagai, M.; Nagai, Y. Tetrahedron Lett. 1974, 4005.
- 14. For a preliminary account see: Fleming, F. F.; Gudipati, V.; Steward, O. W. Org. Lett. 2002, 4, 659.
- 15. Matsumura, K.; Saraie, T.; Hashimoto, N. J. Takeda Res. Lab. 1971, 30, 682.
- 16. Column, or radial, chromatography of 4a results in irreversable absorption on the silica gel.
- 17. Presumbly the electrophilicity of the alkynenitrile is enhanced with an adjacent hydroxyl group whereas the increased steric

demand in the THP-protected alkynenitrile, or locating the hydroxyl group one carbon removed as with 4b, is insufficient to activate the alkynenitrile. An analogous influence of halogen substituents causes an enhanced acidity in haloalkanenitriles^a and activates haloalkanenitriles toward insertion of zinc.^b (a) Makosza, M.; Judka, M. Chem. Eur. J. 2002, 8, 4234. (b) Knochel, P.; Singer, R. D. Chem. Rev. 1993, 93, 2117.

- 18. Earl, R. A.; Townsend, L. B. I. Org. Synth. 1981, 60, 81.
- 19. Murray, R. E.; Zweifel, G. Synthesis 1980, 150.
- 20. The steric demand of *t*-BuMgCl allows deprotonation without premature conjugate addition and, for alkenenitriles, can be substituted by PhMgCl.⁷
- 21. The stereochemistry of 7a was confirmed by comparison with an authentic sample^a whereas $7b-7n$ exhibit analogous chemical shifts that are diagnostic^b for E -alkenenitriles. (a) Fleming, F. F.; Wang, Q.; Steward, O. W. J. Org. Chem. 2001, 66, 2171. (b) Mikolajczak, K. L.; Weisleder, D. Lipids 1978, 13, 514.
- 22. Kim, S. H.; Lee, S. H.; Kang, S. H. Tetrahedron Lett. 1998, 39, 9693.
- 23. Istuno, S.; Darling, G. D.; Lu, P. Z.; Frechet, J. M. J. Polym. Mater. Sci. Engng 1987, 57, 570.
- 24. Ref. 6c. Presumably the nitrile decreases the nucleophilicity of the organometalic since related alkyl-substituted chelates (6 $CN=Me$) do react with aldehydes at ambient temperature: Forgione, P.; Fallis, A. G. Tetrahedron Lett. 2000, 41, 11.
- 25. For related silicate intermediates see: (a) Takeda, K.; Yamawaki, K.; Hatakeyama, N. J. Org. Chem. 2002, 67, 1786. (b) Taguchi, H.; Ghoroku, K.; Tadaki, M.; Tsubouchi, A.; Takeda, T. Org. Lett. 2001, 3, 3811.
- 26. Analogous nitrile-substituted vinyl Grignard reagents alkylate reactive electrophiles as do oxomagnesium chelates without nitrile substituents. 24 Presumably the additive effect of these two deactivating influences is sufficient to prevent 6 from analogous electrophilic alkylations. Thibonnet, J.; Knochel, P. Tetrahedron Lett. 2000, 41, 3319.
- 27. A concerted addition requires simultaneous overlap of the C–Mg bond with the α and β carbons which have opposite phases.
- 28. α -Lithiated alkenenitriles are configurationally unstable: Schmidt, R. R.; Hirsenkorn, R. Tetrahedron 1983, 39, 2043.
- 29. Analogous chelates are proposed as key intermediates in the carbometalation of alkynols.^{6a}
- 30. A related complex is formed between t -BuC \equiv CCH $=$ CHCN and $LiCuMe₂^a$ although $LiCuMe₂$ is generally unreactive toward alkyl-substituted alkenenitriles.⁸ (a) Krause, N.; Wagner, R.; Gerold, A. J. Am. Chem. Soc. 1994, 116, 381.
- 31. For internal complexation of Grignard reagents with π -bonds see: (a) Chinkov, N.; Morlender-Vais, N.; Marek, I. Tetrahedron Lett. 2002, 43, 6009. (b) Hill, E. A. J. Organomet. Chem. 1975, 91, 123.
- 32. (a) Inoue, A.; Kondo, J.; Shinokubo, H.; Oshima, K. Chem. Eur. J. 2002, 8, 1730. (b) Inoue, A.; Kitagawa, K.; Shinokubo, H.; Oshima, K. J. Org. Chem. 2001, 66, 4333.
- 33. X-ray crystallography of 7o confirmed the stereochemical assignment with the Z-stereochemistry of 7p–7t being assigned by analogy. The authors have deposited the crystallographic data for 7o with the Cambridge Crystallographic Data Center (CCDC 177267). The data can be obtained, on request, from the Director, Cambridge Crystallographic Data Center, 12 Union Road, Cambridge, CB2 1EZ, UK.
- 34. No alkylation was observed with MeI, allyl bromide, TMSCl,

PhCOCl or MeOCOCl although alkylation of related a-magnesioalkenenitrile is possible by conversion to the corresponding cuprate.^{[26a](#page-7-0)}

- 35. For general experimental procedures see: Fleming, F. F.; Hussain, Z.; Weaver, D.; Norman, R. E. J. Org. Chem. 1997, 62, 1305. The high-resolution mass spectra were recorded at The Ohio State University Campus Chemical Instrumentation Center.
- 36. Commercially available from Aldrich Chemical Co.
- 37. Landor, S. R.; Demetrion, B.; Grzeskowiak, R.; Pavey, D. F. J. Organomet. Chem. 1975, 93, 129.
- 38. An unusually high $C \equiv N$ frequency and minor discrepancies in NMR shifts were previously reported for nitrile 7e: Tanyeli, C.; Demir, A. S.; Akhmedov, I. M.; Ozgiil, E.; Kandemir, C. G. Synth. Commun. 1996, 26, 2967.
- 39. Prepared by reacting 1.5 equiv. of phenylacetylene with 1.5 equiv. of MeMgCl at 0° C for 15 min.
- 40. Prepared by reacting (Bu)3SnH (1.3 equiv.) with MeMgCl $(1.3$ equiv.) at 0°C for 10 min.
- 41. Bernady, K. F.; Poletto, J. F.; Nocera, J.; Mirando, P.; Schaub, R. E.; Weiss, M. J. J. Org. Chem. 1980, 45, 4702.